

## isotopomer

Isomers having the same number of each isotopic atom but differing in their positions. The term is a contraction of 'isotopic isomer'. Isotopomers can be either constitutional isomers (e.g.  $\text{CH}_2\text{DCH}=\text{O}$  and  $\text{CH}_3\text{CD}=\text{O}$ ) or isotopic stereoisomers [e.g. (*R*)- and (*S*)- $\text{CH}_3\text{CHDOH}$  or (*Z*)- and (*E*)- $\text{CH}_3\text{CH}=\text{CHD}$ ].

### **Source:**

PAC, 1994, 66, 1077 (*Glossary of terms used in physical organic chemistry (IUPAC Recommendations 1994)*) on page 1132

PAC, 1996, 68, 2193 (*Basic terminology of stereochemistry (IUPAC Recommendations 1996)*) on page 2211